

## Supporting Information

### Single-molecule force spectroscopy of an artificial DNA duplex comprising a silver(I)-mediated base pair

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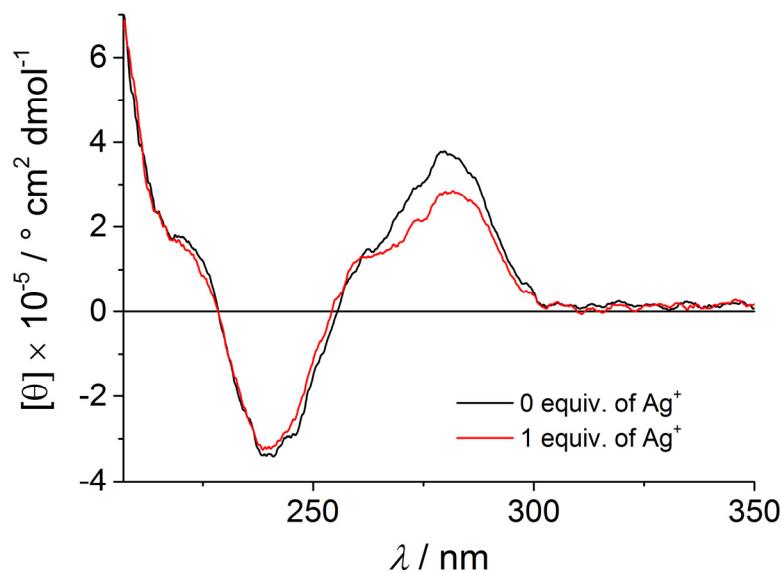
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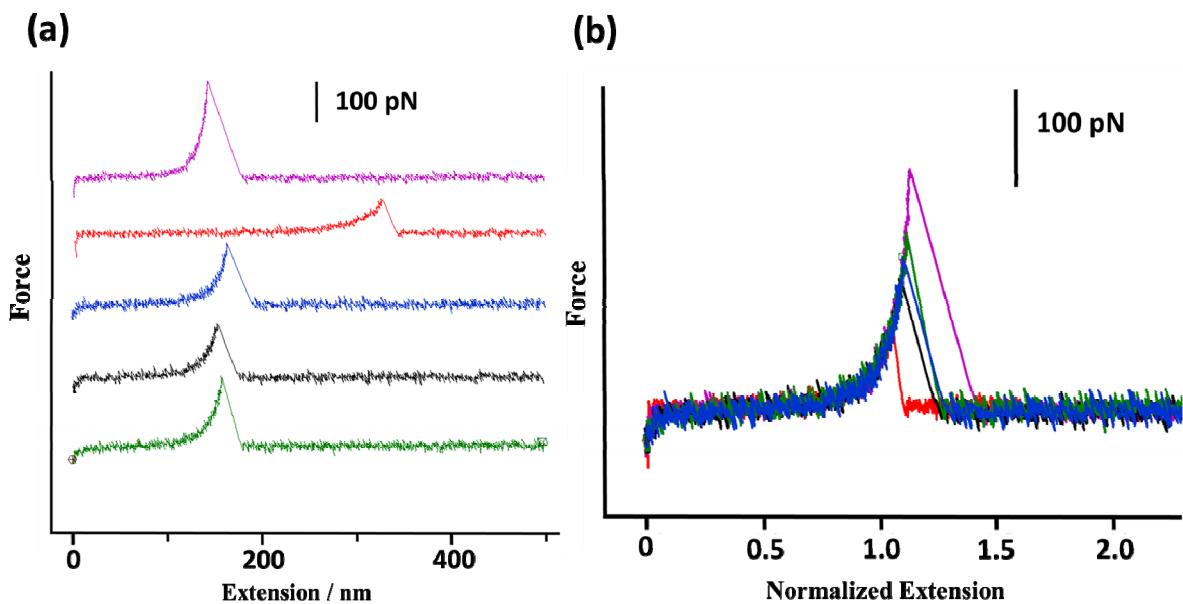
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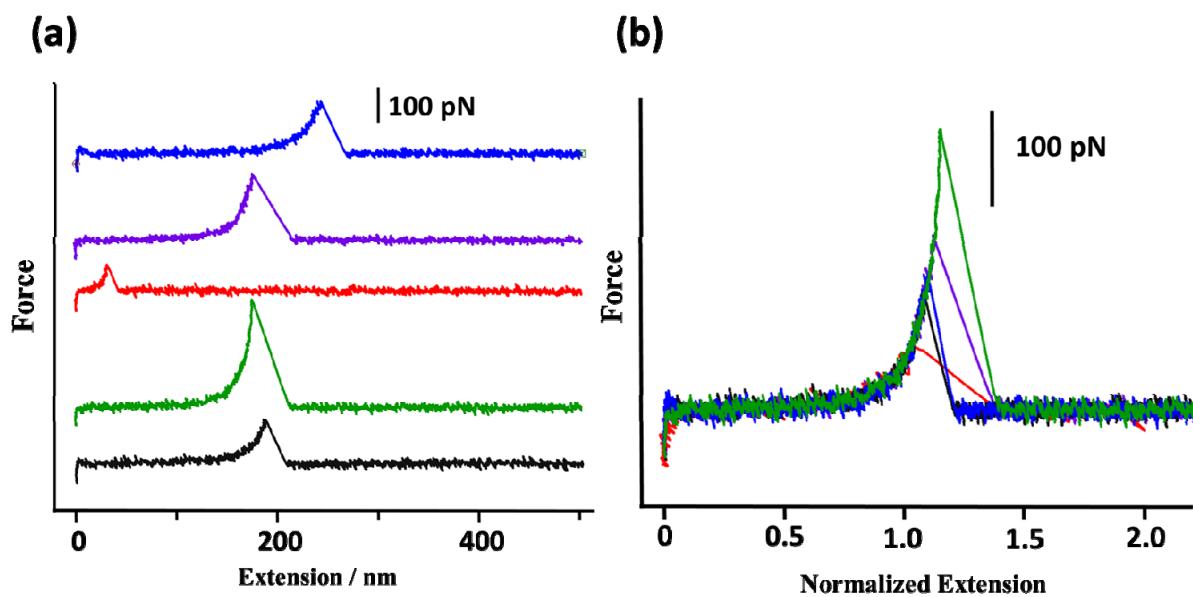


**Figure S1.** CD spectra of duplex **II** in the absence and presence of one equivalent of  $\text{AgNO}_3$ .

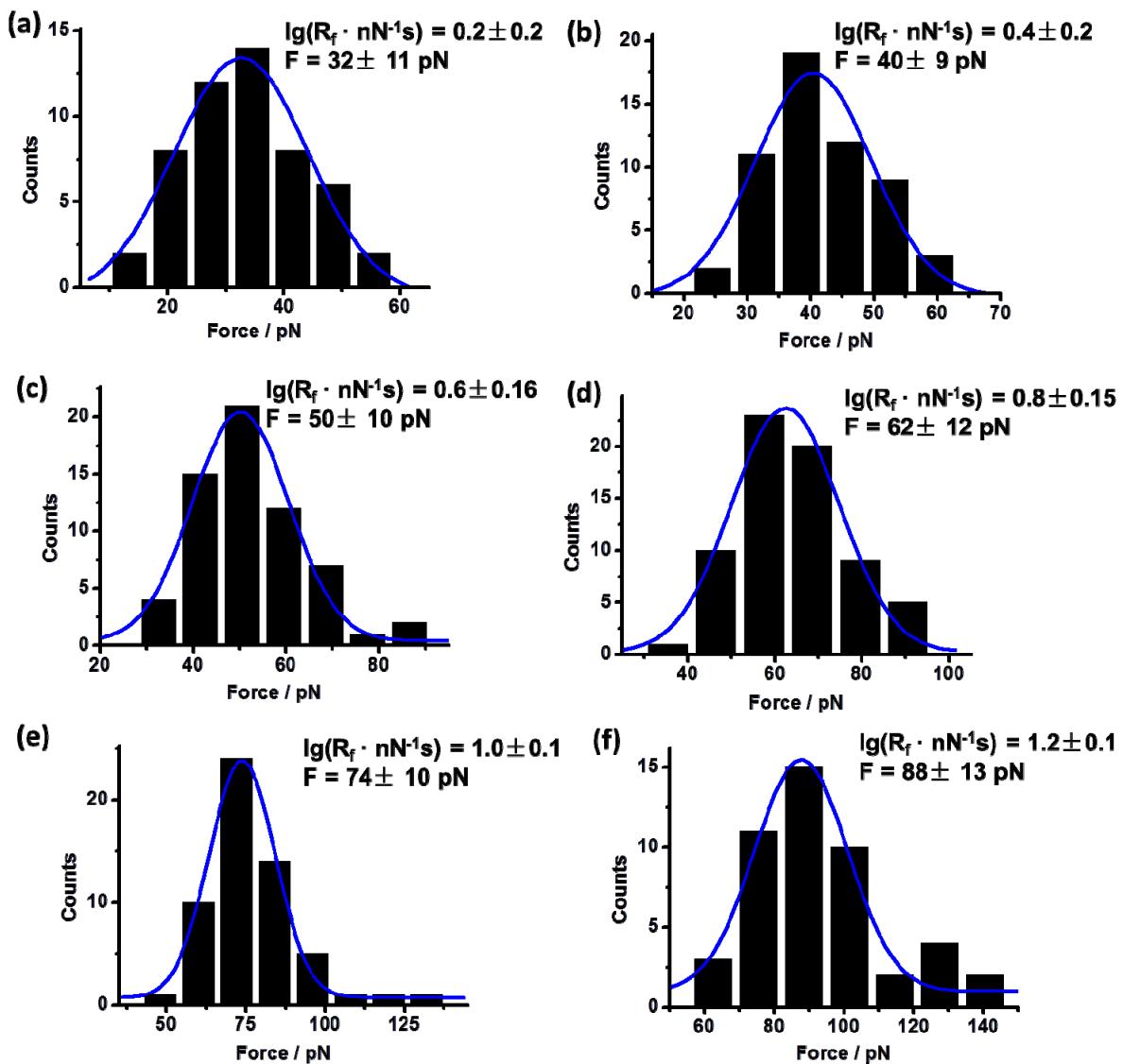
In both cases, the duplex adopts a B-DNA conformation.<sup>1</sup>



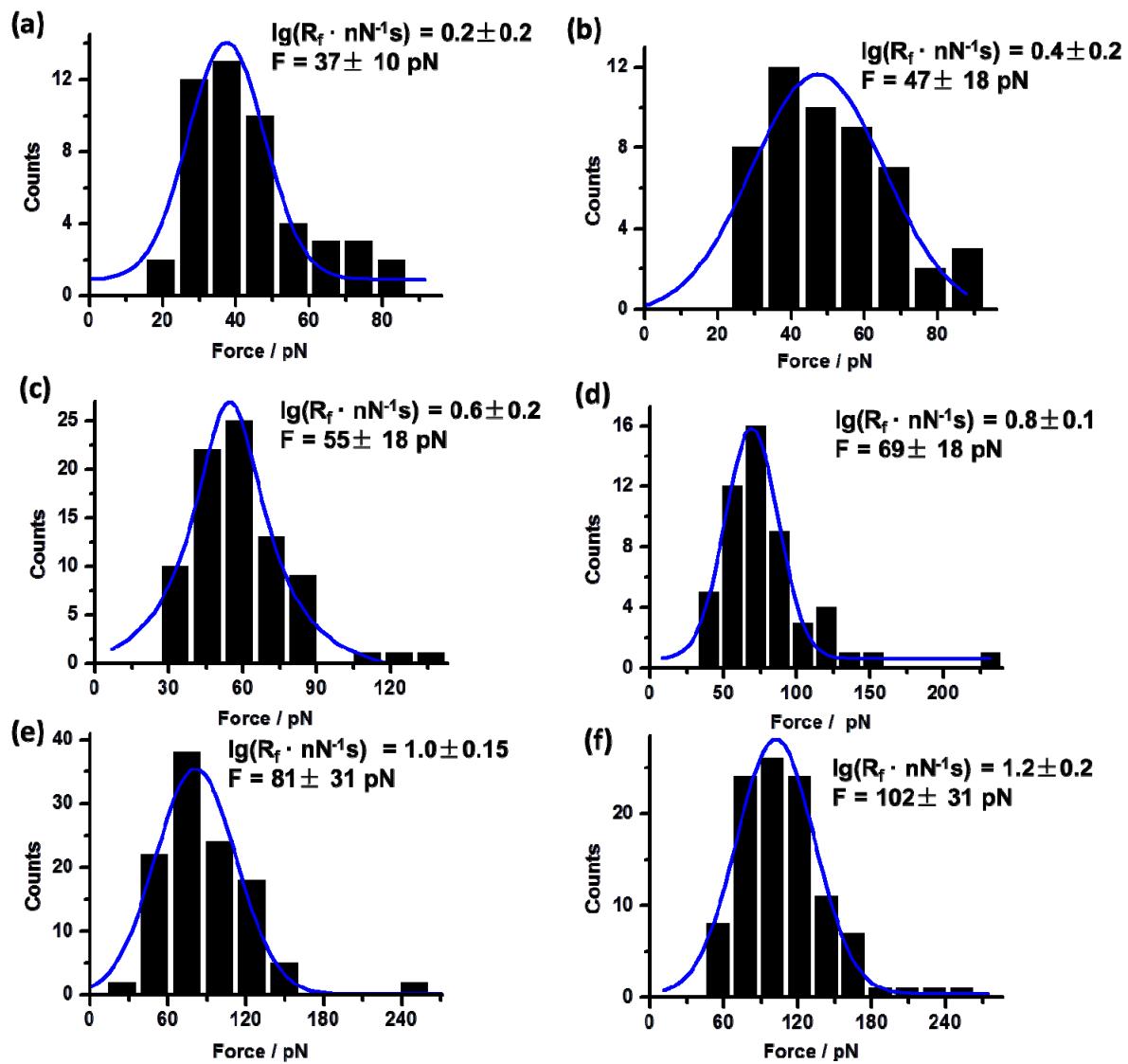
**Figure S2.** (a) Typical force curves obtained from the measurement of duplex **III**. (b) All the normalized force curves are superimposed well.



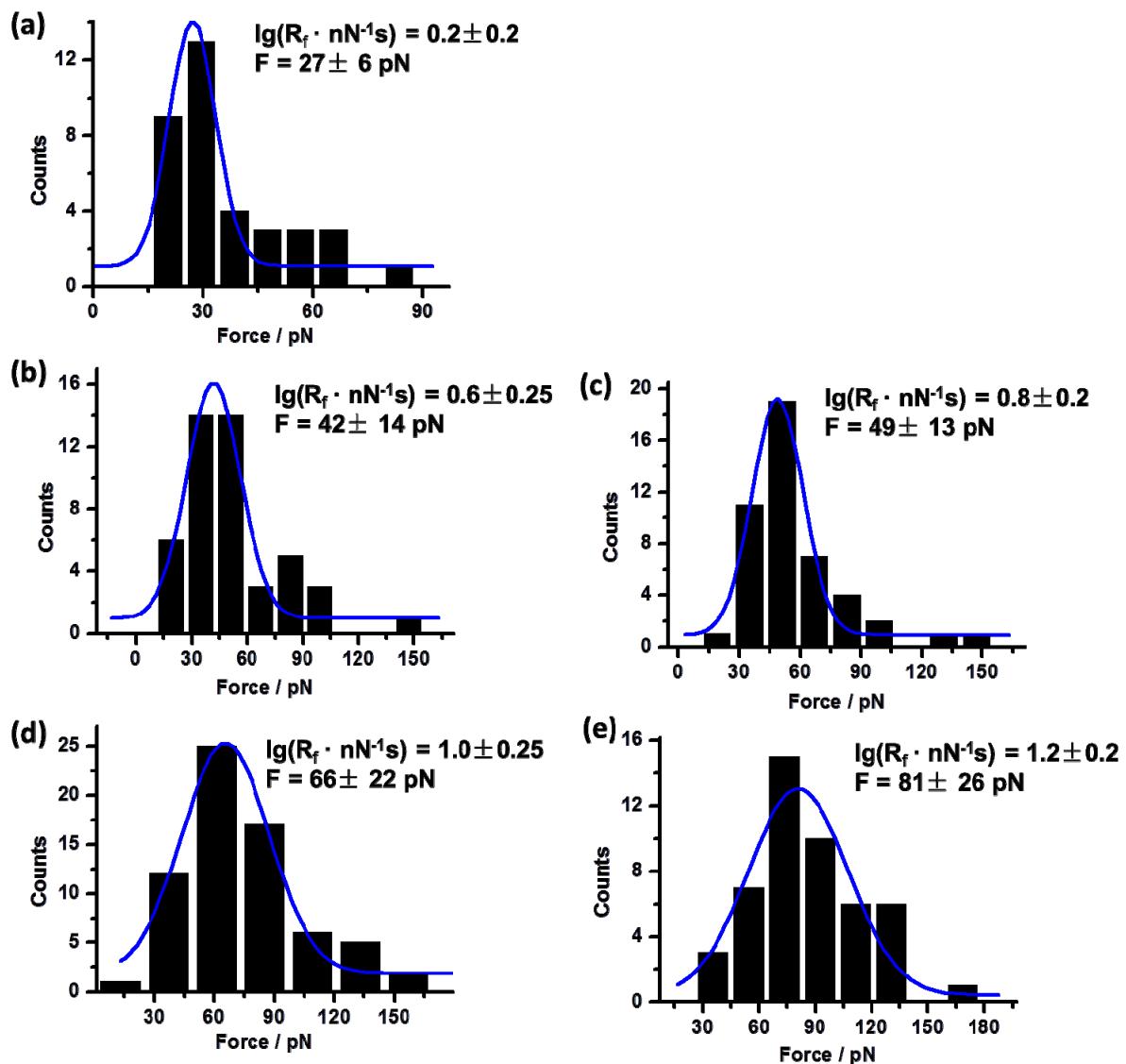
**Figure S3.** (a) Typical force curves obtained from the measurement of duplex **IV** in the absence of  $\text{Ag}^+$ . (b) All the normalized force curves are superimposed well.



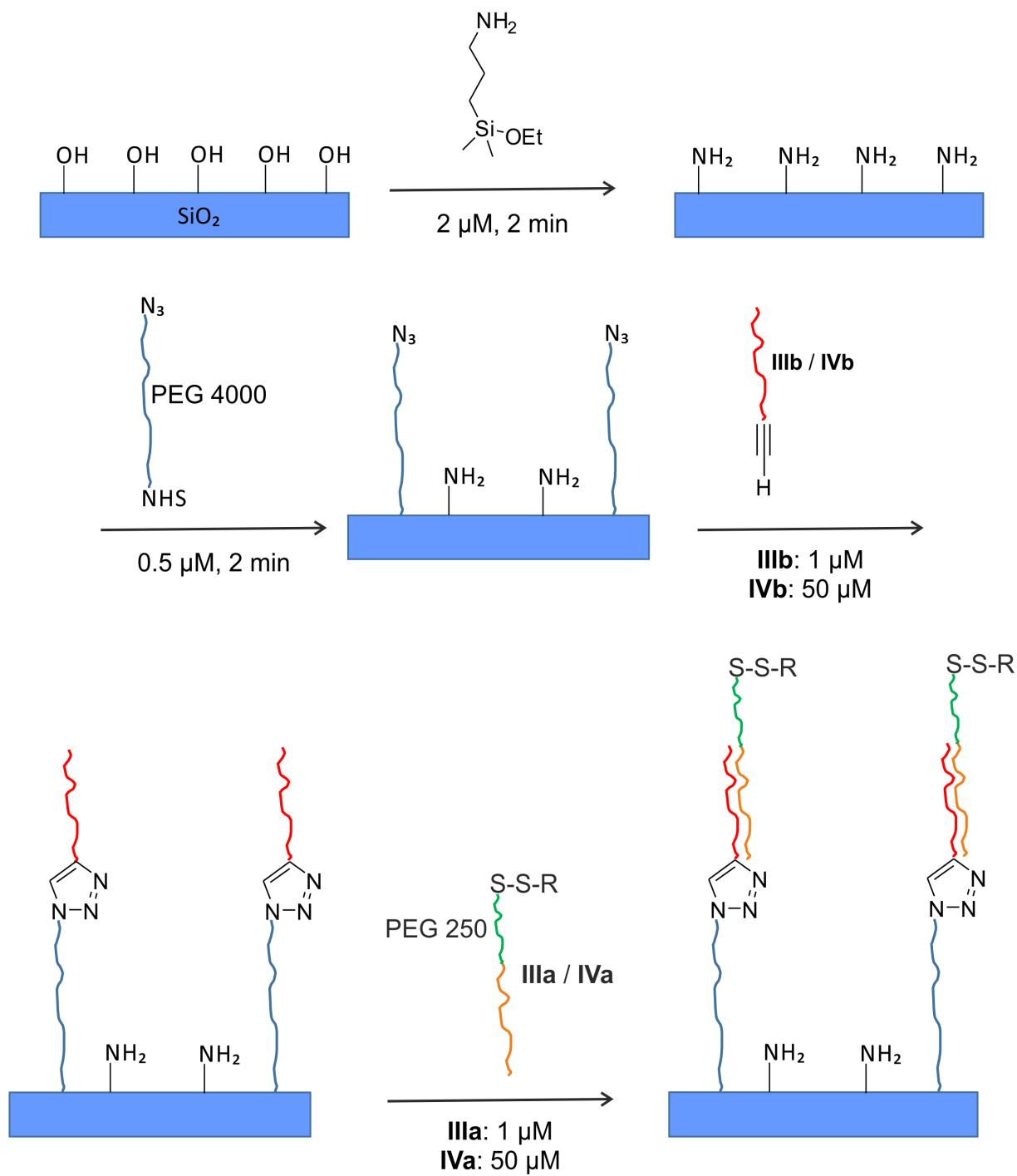
**Figure S4.** The most probable rupture forces of duplex **III** under different  $R_f$ . For building up histogram a, b, c, d, e, f, the number of force curves is 52, 56, 62, 68, 57, and 47, respectively.



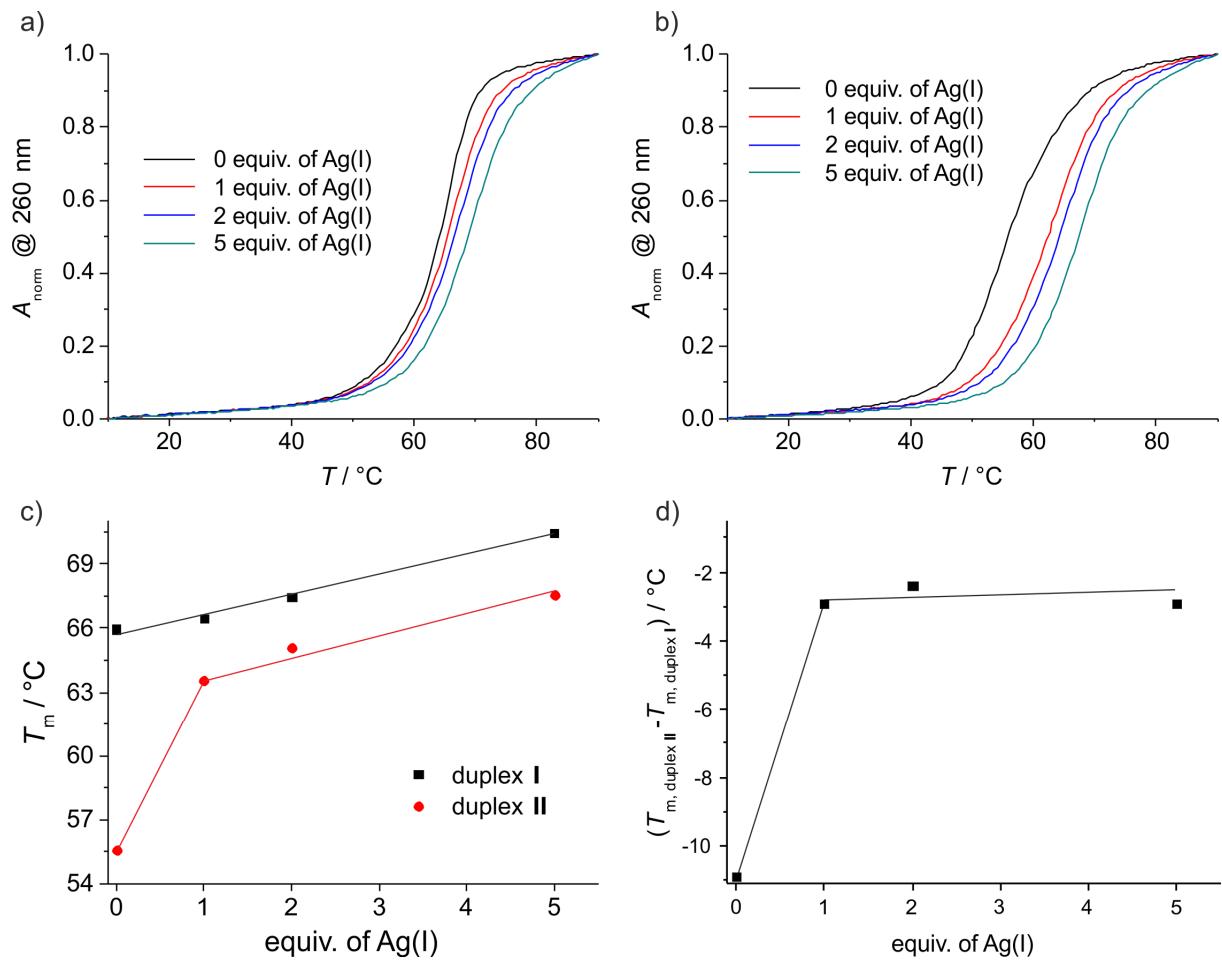
**Figure S5.** The most probable rupture forces of duplex **IV** in the presence of  $\text{Ag}^+$  under different  $R_f$ . For building up histogram a, b, c, d, e, f, the number of force curves is 49, 51, 82, 52, 110, and 104, respectively.



**Figure S6.** The most probable rupture forces of duplex **IV** in the absence of  $\text{Ag}^+$  under different  $R_f$ . For building up histogram a, b, c, d, e, the number of force curves is 35, 46, 46, 68, and 48, respectively.

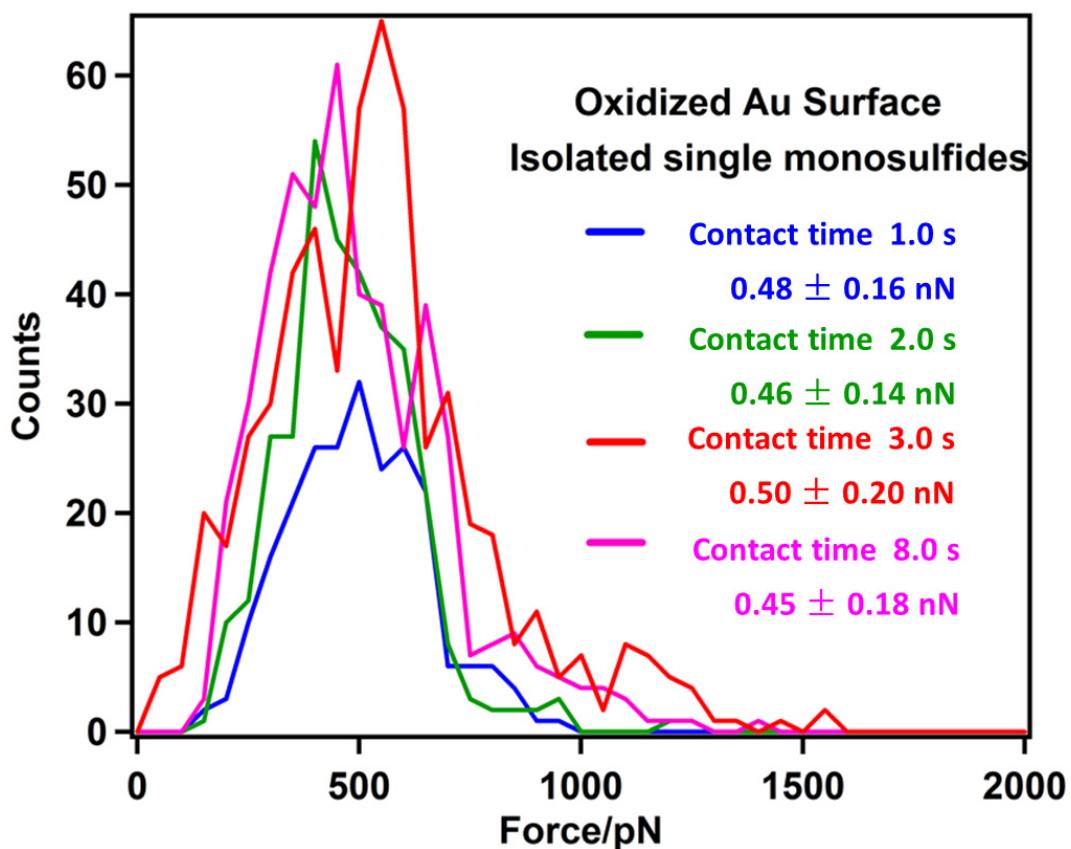


**Figure S7.** Stepwise modification of the SiO<sub>2</sub> surface with oligonucleotide duplexes **III** and **IV**, respectively.



**Figure S8.** a) UV melting curves of duplex **I** in the presence of increasing amounts of Ag(I); b) UV melting curves of duplex **II** in the presence of increasing amounts of Ag(I); c) melting temperatures as derived from the respective UV melting curves, clearly showing the significant stabilization of duplex **II** upon the addition of the first equivalent of Ag(I); d) differences in the melting temperatures of duplexes **II** and **I** upon the addition of increasing amounts of Ag(I) (identical to inset in Figure 2).

In the SMFS experiments, the gold-coated AFM cantilevers were kept into contact with the substrates for 1 s. Here, Au-S coordination bond or even a covalent bond could form. According to the reference, the most probable rupture forces of the Au-S bond with different interaction times are 0.69 nN at 1.0 s, 0.76 nN at 2.0 s, 1.09 nN at 3.0 s and 1.01 nN at 8.0 s, respectively, indicating the shifting of bond types from coordinate to covalent.<sup>2</sup> Even in the isolated single monosulfides system that can form only Au-S coordination bond, the most probable rupture force is larger than 450 pN, and the force value is independent on the interaction time, as shown in Figure S9. Therefore, the Au-S bond is much stronger than the forces holding together the DNA duplex, and the weakest part of the polymer bridge is the DNA duplex.



**Figure S9.** The most probable rupture force of Au-S coordination bond is larger than 450 pN, and the force value is independent on the interaction time.

**Table S1.** Modification of SiO<sub>2</sub> substrates characterized by XPS.

Atom	C <sub>1s</sub>	O <sub>1s</sub>	Si <sub>2p</sub>	N <sub>1s</sub>	P <sub>2p</sub>	S <sub>2p</sub>	Ag <sub>3d</sub>
%							
Before step 1	16.46	60.57	22.97	-	-	-	-
%							
After step 1	15.33	1.96	60.36	22.34	-	-	-
%							
After step 2	24.59	51.91	20.71	2.79	-	-	-
%							
After step 3							
<b>III b</b>	20.00	56.86	20.23	2.58	0.34	-	-
<b>IV b</b>	22.14	55.63	19.12	2.74	0.38	-	-
%							
After step 4							
duplex <b>III</b>	23.89	54.32	17.23	3.14	0.36	1.06	-
duplex <b>IV</b>	31.89	43.42	16.31	4.88	0.31	3.20	-
duplex <b>IV</b> in the presence of Ag <sup>+</sup>	31.01	43.33	15.13	5.11	0.84	1.98	2.60

## References

- 1 Vorlíčková, M.; Kejnovská, I.; Bednářová, K.; Renčíuk, D.; Kypr, J. Circular dichroism spectroscopy of DNA: from duplexes to quadruplexes. *Chirality* **2012**, *24*, 691–698.
- 2 Xue, Y.; Li, X.; Li, H.; Zhang, W. Quantifying thiol–gold interactions towards the efficient strength control. *Nat. Commun.* **2014**, *5*, 4348.