

# The Cahn-Hilliard Equation

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The Cahn-Hilliard equation describes the process of spinodal decomposition of a binary mixture.

We consider a binary mixture of the two components  $A$  and  $B$ , which are described by their local per-volume densities  $c_A(\mathbf{x}, t)$  and  $c_B(\mathbf{x}, t)$ , respectively. For a binary mixture, we can assume that  $c_A(\mathbf{x}, t) + c_B(\mathbf{x}, t) = 1$ , and therefore only use one concentration  $c(\mathbf{x}, t)$  for the description:

$$c_A(\mathbf{x}, t) := c(\mathbf{x}, t) \quad \text{and} \quad c_B(\mathbf{x}, t) = 1 - c(\mathbf{x}, t) \quad (1)$$

The corresponding flux can be determined to be

$$\mathbf{J} = -M \nabla (\mu_B - \mu_A), \quad (2)$$

where  $M$  is a mobility coefficient and  $\mu_i$  is the chemical potential of component  $i$ .

According to classical thermodynamics, the difference of the two chemical potentials  $\mu_B - \mu_A$  can be expressed in terms of a variation of a corresponding free energy potential  $\mathcal{F}[c]$ :

$$\mu_B - \mu_A = \frac{\delta \mathcal{F}[c]}{\delta c}, \quad (3)$$

$$\Rightarrow \mathbf{J} = -M \nabla \frac{\delta \mathcal{F}[c]}{\delta c}. \quad (4)$$

The assumption of mass conservation directly leads to the Cahn-Hilliard equation:

$$\frac{\partial}{\partial t} c(\mathbf{x}, t) = -\nabla \cdot \mathbf{J} \quad (5)$$

$$= \nabla \cdot M \nabla \frac{\delta \mathcal{F}[c]}{\delta c}. \quad (6)$$

A typical choice for the free energy is

$$\mathcal{F}[c] = \int_{\Omega} \frac{\kappa}{2} |\nabla c|^2 + f(c) \, d\mathbf{x}, \quad (7)$$

with a double well potential for the local energy  $f(c)$

$$f(c) = \frac{a}{2} c^2 + \frac{b}{4} c^4 \quad (8)$$

$$\Rightarrow \frac{\partial}{\partial t} c(\mathbf{x}, t) = \nabla \cdot M \nabla (-\kappa \Delta c(\mathbf{x}, t) + ac(\mathbf{x}, t) + bc(\mathbf{x}, t)^3). \quad (9)$$

The coefficient  $a$  is related to the temperature  $T$  of the binary mixture and decides, whether a phase separation will occur. A typical phase diagram for such a system is shown in Fig. 1

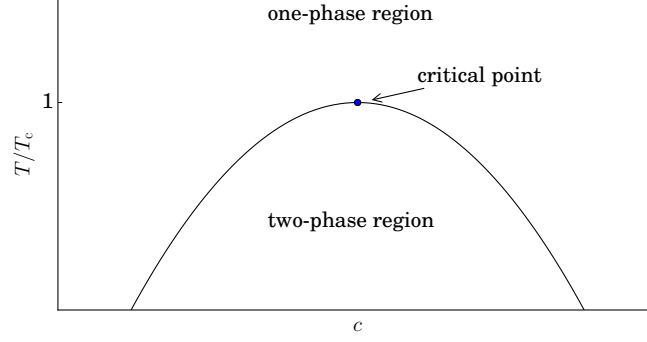


Figure 1: typical phase diagram of a binary mixture.

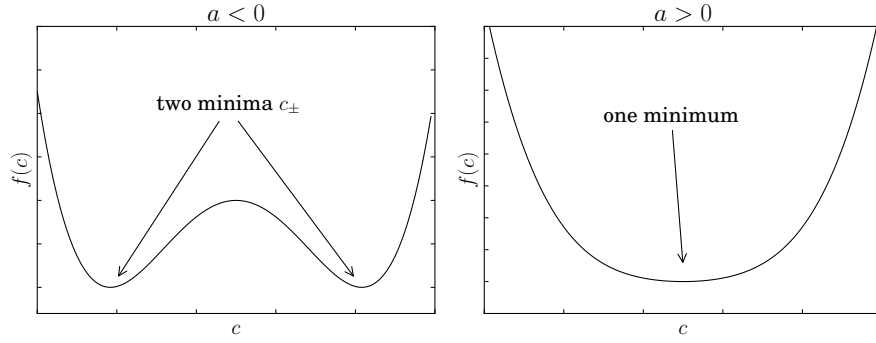


Figure 2: Local free energy  $f(x)$  of a binary mixture below the critical temperature ( $T < T_{\text{crit}}$ ,  $a < 0$ , left) and above the critical temperature ( $T > T_{\text{crit}}$ ,  $a > 0$ ). For  $T < T_{\text{crit}}$ , the free energy has two local minima  $c_{\pm}$ , which merge into only one minimum for  $T \geq T_{\text{crit}}$ .

## 0.1 Linear Stability Analysis

To analyze the linear stability of the Cahn-Hilliard equation, we use as an ansatz a homogeneous solution  $c_0$  with a perturbation with small amplitude  $\epsilon$ , growth rate  $\beta$  and spatial wavenumber  $k$ :

$$c(\mathbf{x}, t) = c_0 + \epsilon e^{\beta t} e^{i k x} \quad (10)$$

Inserting this ansatz in the Cahn-Hilliard equation and omitting all terms of  $\mathcal{O}(\epsilon^2)$  and higher, one obtains:

$$\beta(k) = M(-k^2)(\kappa k^4 + a + 3bc_0^2) \quad (11)$$

$$= \underbrace{-M\kappa k^4}_{<0} \underbrace{-M(a + 3bc_0^2)k^2}_{\leq 0, \text{ depending on } a, b, c_0} \quad (12)$$

Therefore the stability depends on the particular choice of  $a$ ,  $b$  and  $c_0$ . The fastest growing mode can be determined by finding the local maximum  $k_{\max}$  of the dispersion relation  $\beta(k)$ :

$$\left. \frac{\partial \beta}{\partial k} \right|_{k=k_{\max}} = 0 \quad (13)$$

$$\Leftrightarrow k_{\max} = \sqrt{\frac{a + 3bc_0}{2\kappa}} \quad (14)$$

$$(15)$$

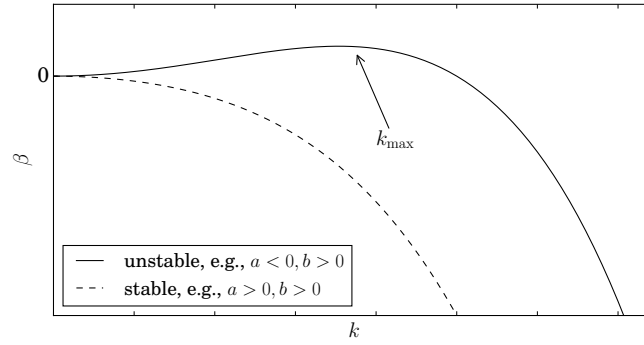


Figure 3: Growth rate  $\beta$  of a perturbation with wavenumber  $k$  determined by linear stability analysis.

## 0.2 Long-term behavior

Just as with the Allen-Cahn equation, the dynamics of the Cahn-Hilliard equation can only decrease the free energy  $\mathcal{F}[c]$  over time:

$$\frac{d}{dt}\mathcal{F}[c] = \int_{\Omega} \frac{\delta\mathcal{F}}{\delta c} \frac{\partial c}{\partial t} d\mathbf{x} \quad (16)$$

$$\stackrel{\text{C-H eq.}}{=} \int_{\Omega} \frac{\delta\mathcal{F}}{\delta c} \nabla \cdot M \nabla \frac{\delta\mathcal{F}[c]}{\delta c} d\mathbf{x} \quad (17)$$

$$\stackrel{\text{int. by parts}}{=} - \int_{\Omega} \nabla \frac{\delta\mathcal{F}}{\delta c} \cdot M \nabla \frac{\delta\mathcal{F}[c]}{\delta c} d\mathbf{x} \quad (18)$$

$$= - \int_{\Omega} \underbrace{M}_{>0} \underbrace{\left( \nabla \frac{\delta\mathcal{F}[c]}{\delta c} \right)^2}_{\geq 0} d\mathbf{x} \leq 0. \quad (19)$$

The minimization of  $\mathcal{F}$  can be achieved in two ways:

- a) minimize  $f(c)$ ,
  - b) minimize  $\frac{\kappa}{2} |\nabla c|^2$ .
- a) After a short time, almost everywhere  $c = c_+$  or  $c = c_-$ , as  $f(c)$  is minimal there.
- b)  $|\nabla c|^2$  is an energetic penalty for gradients of  $c$ , therefore the dynamics tend to reduce boundaries between regions with  $c = c_+$  or  $c = c_-$ . This leads to coarsening dynamics, where regions with equal concentration  $c$  merge to larger regions with less boundaries.

One can show that the typical length scale  $l$  of the observable pattern evolves as

$$l \sim t^{1/3}, \quad (20)$$

i.e., the coarsening process is very slow.