

KPFM investigations on polysiloxane coated gold and glass substrates using pure and functionalized cantilevers



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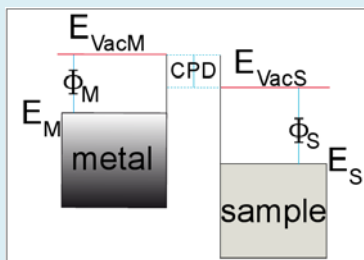
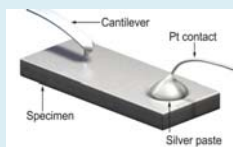
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Atomic force microscopy (AFM) measurements provide new insights into the working of battery materials on the micro- and nanoscale. Kelvin Probe Force Microscopy (KPFM) [1] is an AFM mode, where in addition to the sample topography the surface potential of the material can be mapped. On hard substrates this imaging mode can normally be executed quite easily, but sticky samples such as polymers can provide an obstacle.

This can be avoided by using specially functionalized cantilevers, which do not stick to the surface. A model experiment setup was used to gather information on the effect of this functionalization on the measured surface potential, and to determine if KPFM can be a valuable additional method for research on polymer electrolytes or polymer coated electrodes in lithium polymer batteries, e.g. in investigations concerning the ageing of battery materials[2].

KPFM measurements

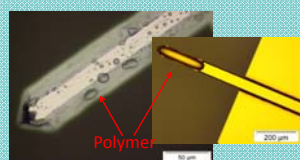
The Kelvin Probe Force Microscopy mode enables measurement of the sample surface potential (SP) and yields information on work function of a surface.



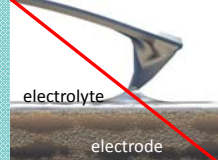
Background

Problem: soft polymer samples contaminate AFM levers and make AFM measurements impossible

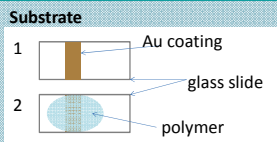
Aim: avoiding a polymer meniscus during measurements, electrochemical measurements still possible



Polymer repellent coating



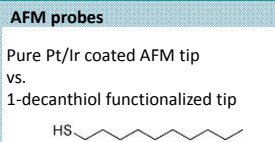
Samples



Preparation:
Preparing glass slides with gold coating by evaporation method

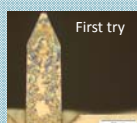
Polymer: mixture of epopolysiloxane and MEEP-cyclophosphazene (1:1) with DMEDA as crosslinker [3]

Preparing polymer thin films with doctor blade method and crosslinking at 70 °C

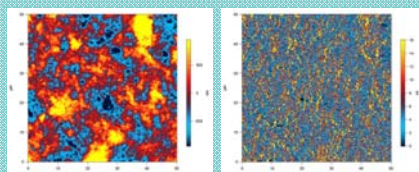


Preparation:
Cleaning of tips by immersing in conc. H₂O₂ for 30 s and then in 0.1 M H₂O₂ + 0.1 M HCl overnight

Coating by immersing in 0.05 mMol decanthal solution in chloroform over night



Why not take 'real' batteries for comparison?



- heterogenous conductivity and surface potential
- relatively high roughness leads to high amount of artefacts in images
- sensitive to humidity and oxygen

Topography and conductivity image of electrode

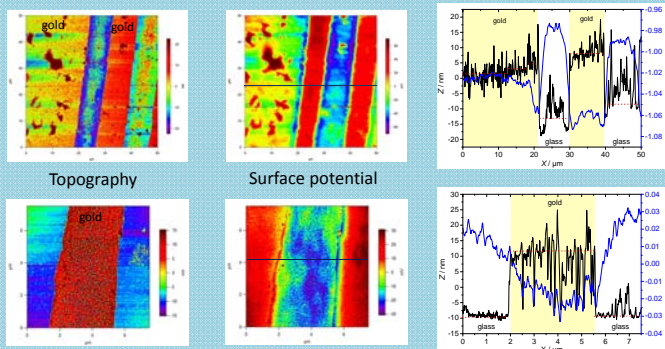
Conclusion:

It was shown, that with the KPFM mode it is possible to distinguish very well between a conductive material and an isolator, and even to measure layer thicknesses of a known material.

Functionalized cantilevers can in principle be used for KPFM measurements, although in this experimental setup the layer thickness of the polymer was too high to resolve the underlying features even with pure cantilevers. Nevertheless, KPFM can be a useful tool to improve the knowledge of the processes going on in batteries at the micro- or nanoscale [2].

Results

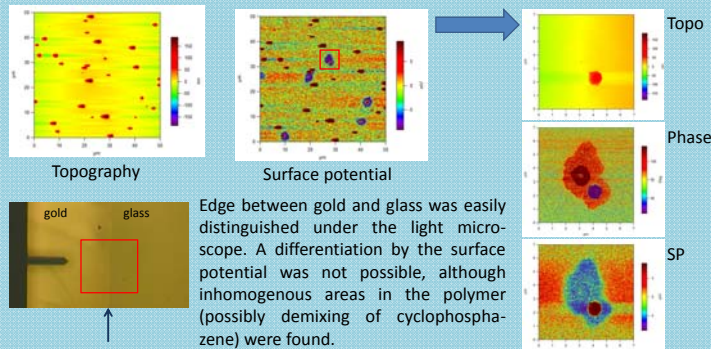
Surface potential difference between glass and gold measured with pure cantilevers



Potential difference between glass and gold is ca. 50 mV. A strong dependence between the SP signal and the gold layer thickness was observed.

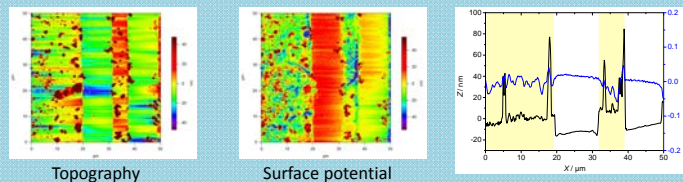


Surface potential difference with polymer coating measured with pure cantilevers

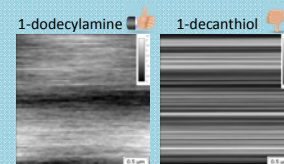


Edge between gold and glass was easily distinguished under the light microscope. A differentiation by the surface potential was not possible, although inhomogeneous areas in the polymer (possibly demixing of cyclophosphazene) were found.

Surface potential difference between glass and gold with functionalized cantilevers (no polymer coating on the sample)



In an additional test with non-crosslinked polysiloxane on a glass substrate, the quality of the topography images was higher if functionalized cantilevers were used. A functionalization with 1-dodecylamine on Si₃N₄ tips showed better results than 1-decanthiol, though, but these tips cannot be applied for KPFM. Therefore, the application of functionalized levers for KPFM imaging of sticky samples is feasible to improve the measurements.



Test of stickiness

Literature:

- [1] Nonnenmacher et al., *Appl. Phys. Lett.* (1991) **58**: 2921-2923.
- [2] Kalinin, Balke, *Adv. Mater.* (2010) **22**: E193-209
- [3] Z. Phys. Chemie-Int. J. Res. Phys. Chem. Chem. Phys., **224** (2010) 1439-1473

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